A High Potential Optical Method for Sensitive and Selective Detection of Essential Elements

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Abstract: An optical sensor based on surface plasmon resonance (SPR) phenomenon has been emerged for sensing a variety of analytes, including essential elements. It has numerous advantages, which include very high sensitivity, simple sample preparation, low cost, fast measurement capability, no requirement for reference solution, high reproducibility, ability to monitor kinetic behaviour, label-free detection, and nondestructive. The instrument and different developments on active layers or recognition molecules for sensitivity and selectivity improvements were explored. Further investigation in SPR optical sensor technology will expand SPR detection abilities and allow SPR sensing to be used widely including in environmental monitoring as an effective essential elements sensor in the future.

Keywords: Surface plasmon resonance, essential elements, sensitive, optical sensor

INTRODUCTION

Numerous SPR sensing structures for chemical and biochemical sensing have been developing tremendously for the past decades. Since the beginning of 21st century, SPR for sensing essential elements has been started to receive continuously growing attention from scientific community. In this study, the development of SPR for essential elements detection by our research group as well as other research groups all over the world has been explored and discussed.

MATERIALS AND METHODS

Surface plasmon requires a coupler as it does not couple with the outside electromagnetic radiation. The simplest coupler for surface plasmon excitation is the prism coupler. Figure 1 shows the schematic diagram for SPR setup. The Kretschmann configuration is commonly used because it is the most efficient. A metal film was placed directly on the horizontal surface of the film (~50 nm) and the incident beam passed through the film to excite the surface plasmon at the bottom of the film. A lock-in amplifier and chopper were used in the setup to enhance the signal-to-noise-ratio of the measurement, while a filter was used to modulate the intensity of the incident light.

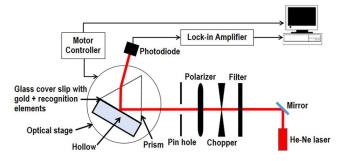


Fig. 1. SPR experimental setup.

RESULTS AND DISCUSSION

The most important finding for sensing essential elements application in SPR is the modification of gold surface with recognition molecules for particular essential elements. Various recognition elements for particular detection is shown in Table 1.

Table 1. Recognition elements in SPR for sensing particular essential elements.		
Detection limit/range	Active layers/recognition elements	Reference
1.0 pM - 10 mM	Squarylium dye containing polymeric thin film	1
2.0 μM-1.0 mM	Immobilize methallothionein onto a carboxymethylated	2
	dextran matrix	
Zn ²⁺ , Ni ²⁺ 10 nM	Immobilize bovine serum albumin onto a	3
	carboxymethylated dextran matrix	
Cu ²⁺ : 1.6 nM	Cu ²⁺ : NH ₂ – Gly – Gly – His – COOH	4
Ni ²⁺ : 41 pM	Ni ²⁺ : NH ₂ – (His) ₆ – COOH	
Cu ²⁺ 0.1 µM-1.0 mM (selective)	2-aminoethane thiolhydrochloride and 6-aminohexane	5
	thiolhydrochloride	
1pM- 10 mM	Calix[4]crown-5 derivatives	6
0.2 mM - 5.0 mM	Gold nanoparticles-chitosan composite	7
7.9 μM - 1.6 mM	MMW chitosan (glutaraldehyde-crosslinked)	8
~8.0 μM-2.0mM	MMW chitosan (glutaraldehyde-crosslinked)	9
Zn ²⁺ : 15 μM; Ni ²⁺ : 17 μM	polypyrrole-chitosan	10
0.85 nM	N-[5-(3'-maleimidopropylamido)-1-	11
	carboxypentyl]iminodiacetic acid (NTA) & polyhistidine	
1.52 μM	chitosan and chitosan-tetrabutylthiuram disulfide	12
1 µM	5-Chloro-2-[(1E,2E)-3-(4	13
·	(dimethylamino)phenyl)allylidene)amino)]phenol	
0.17µM - 1.7 mM	4-(2-pyridylazo) resorcinol-chitosan-graphene oxide	14
25.57 nM	Valinomycin-chitosan-graphene oxide	15
	Detection limit/range 1.0 pM - 10 mM 2.0 μM-1.0 mM 10 nM Cu ²⁺ : 1.6 nM Ni ²⁺ : 41 pM 0.1 μM-1.0 mM (selective) 1pM- 10 mM 0.2 mM - 5.0 mM 7.9 μM - 1.6 mM ~8.0 μM-2.0mM Zn ²⁺ : 15 μM; Ni ²⁺ : 17 μM 0.85 nM 1.52 μM 1 μM 0.17μM - 1.7 mM	Detection limit/rangeActive layers/recognition elements1.0 pM - 10 mMSquarylium dye containing polymeric thin film2.0 μM-1.0 mMImmobilize methallothionein onto a carboxymethylated dextran matrix10 nMImmobilize bovine serum albumin onto a carboxymethylated dextran matrix10 nMImmobilize bovine serum albumin onto a carboxymethylated dextran matrixCu ²⁺ : 1.6 nMCu ²⁺ : NH ₂ – Gly – Gly – His – COOH Ni ²⁺ : 41 pM0.1 μM-1.0 mM2-aminoethane thiolhydrochloride and 6-aminohexane (selective)10 nMCalix[4]crown-5 derivatives0.2 mM - 5.0 mMGold nanoparticles-chitosan composite7.9 μM - 1.6 mMMMW chitosan (glutaraldehyde-crosslinked) MW chitosan (glutaraldehyde-crosslinked)2 ²⁺ : 15 μM; Ni ²⁺ : 17 μMpolypyrrole-chitosan carboxypentyl]iminodiacetic acid (NTA) &polyhistidine chitosan and chitosan-tetrabutylthiuram disulfide1.52 μMchitosan and chitosan-tetrabutylthiuram disulfide thitosan (glutaralopen)]phenol0.17μM - 1.7 mM4-(2-pyridylazo) resorcinol-chitosan-graphene oxide

Table 1. Recognition elements in SPR for sensing particular essential elements.

CONCLUSIONS

The evolution of SPR in essential elements sensing have been explored. Various modifications of the metal surface to increase the sensitivity and selectivity were summarized. SPR has a high sensitivity to sense metal ions as low as pM to μ M. Fast and label-free essential elements can be measured in real time with small sample consumption. Further research will expand the SPR detection abilities and allow SPR sensing to be used widely as an effective essential elements sensor.

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